

MATERIAL SAFETY DATA SHEET

ZINC, GRANULAR

PRODUCT CODE NUMBER(S): 9060-1

PRODUCT IDENTIFICATION

Chemical Name and Synonyms: Zinc, granular, No. 20

mesh

Chemical Family: Metal Chemical Formula: Zn

Product Use: Laboratory reagent
Manufacturer's Name and Address:
Caledon Laboratories Ltd.
40 Armstrong Avenue

Georgetown, Ontario L7G 4R9 **Telephone No:** (905) 877-0101

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HAZARDOUS INGREDIENTS OF MATERIALS

Ingredients % TLV Units CAS No.

Zinc >99 Not established 7440-66-6

PHYSICAL DATA

Physical State: Solid

Odour and Appearance: Bluish-gray metal granules or

shot; odourless

Odour Threshold (ppm): Not applicable
Vapour Pressure (mm Hg): 1 mm Hg at 487°C
Vapour Density (Air = 1): Not applicable
Evaporation Rate: Not applicable

Boiling Point (degrees C): 908°C Melting Point (degrees C): 419.5°C

pH: Not available Specific Gravity: 7.140

Coefficient of Water/Oil distribution: Not available

SHIPPING DESCRIPTION

UN: Not regulated (as granules or shot)

T.D.G. Class: Not regulated Pkg. Group: Not regulated

REACTIVITY DATA

Chemical Stability: Stable in dry air. Becomes covered with a white coating of basic carbonate on exposure to moist air. Damp powder may heat spontaneously and ignite on exposure to air.

Incompatibility with other substances: Incompatible with water, acids, strong oxidizers, halogenated hydrocarbons, strong alkalies, alkali hydroxides, sulphur. Contact with strong acids or alkali hydroxides releases flammable/explosive hydrogen gas. Contact with acidic arsenic solutions may release toxic arsene gas. Reacts violently with sulphur. Reacts explosively with halogenated hydrocarbons if heated.

Powdered zinc becomes incandescent or ignites in the presence of fluorine, bromine, or chlorine. Wetted mixtures with ammonium nitrate react violently. Powder may ignite spontaneously with air, especially when damp. Powder can form explosive mixtures with air and may explode if ignited by flame.

Reactivity: Avoid excessive heat, all ignition sources, generation of dust, all incompatible materials. Avoid water. **Hazardous Decomposition Products:** Explosive hydro-

gen gas, ZnO

FIRE AND EXPLOSION DATA

Flammability: Metal is not combustible. Powder form is flammable under almost all ambient conditions.

Extinguishing Media: USE ONLY dry powder or sand (for powder). Avoid water since contact will cause the liberation of highly flammable gases. Fight fire from safe distance and from upwind. Firefighters must wear protective equipment and clothing sufficient to prevent inhalation of dust or fumes and contact with skin and eyes.

Flash Point (Method Used): Not available

Autoignition Temperature: 650°C (dust cloud), 460°C

(layers)

Upper Flammable Limit (% by volume): Not available

Lower Flammable Limit (% by volume): 0.5g/L

Hazardous Combustion Products: Explosive hydrogen

gas, ZnO

Sensitivity to Impact: None identified

Sensitivity to Static discharge: Mixtures of dust in air may

be ignited by static discharge

TOXICOLOGICAL PROPERTIES AND HEALTH DATA

Toxicological Data:

LD₅₀: Not available **LC**₅₀: Not available

Effects of Acute Exposure to Product:

Inhaled: Inhalation of dust or fumes may cause zinc fume fever, with a metallic taste in the mouth, and flu-like symptoms, including throat dryness, cough, weakness, generalized aching, chills, fever, nausea and vomiting. Symptoms last about 24 hours after exposure, and recovery is complete with nopermanent effects.

In contact with skin: Dust may cause mechanical irritation.
In contact with eyes: Dust may cause mechanical irritation

Ingested: Not an expected route of exposure. Would probably cause gastrointestinal irritation and zinc fume fever, as in "Inhaled", if large amounts were ingested.

Effects of Chronic Exposure to Product:

Carcinogenicity: No evidence of carcinogenicity

Teratogenicity: No information available

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Reproductive Effects: No information available

Mutagenicity: No information available

Synergistic Products: Zinc is antagonistic to the toxic effects of cadmium to the kidneys. Zinc is also antagonistic to

some of the toxic effects of lead.

PREVENTIVE MEASURES

Engineering Controls: Non-sparking, grounded, separate, exhaust ventilation required, if dust is present.

Respiratory Protection: Dust mask if dust is present. NIOSH/MSHA approved self-contained breathing apparatus for high or unknown concentrations, as in fire or spill conditions.

Eye Protection: Chemical safety goggles. Use heat resistant goggles if working with molten form.

Skin Protection: Wear protective gloves and clean body-covering clothing. Wear heat resistant clothing if working with molten form. Zinc dust can react with chlorinated hydrocarbons such as CPE or PVC when heated.

Other Personal Protective Equipment: Safety shower and eye-wash fountain in work area.

Leak and Spill Procedure: If dust is present, eliminate all sources of igntion. In dusty conditions, cleanup personnel must be thoroughly trained in the hazards of this material and must wear protective equipment and clothing sufficient to prevent inhalation of dust or fumes, and contact with skin and eyes. Mix dust with dry sand or other inert material, transfer carefully into container and arrange for collection by disposal company. Granules or wire should be recycled. Wash site of spillage thoroughly with water and detergent.

Waste Disposal: Follow all federal, provincial and local regulations for disposal. Use only licensed disposal and waste hauling companies.

Handling Procedures and Equipment: Workers handling this material must be thoroughly trained in its hazards and its safe use. Use the smallest amount possible for the purpose, in designated areas with adequate ventilation. Follow routine safe handling and good housekeeping procedures. Avoid generating dust. Avoid use of water. Avoid contact with skin and eyes and inhalation of dust. If dust is present, elminate all ignition sources, and uses non-sparking tools.

Storage Requirements: Store in suitable, labelled containers, in a cool, dry, well-ventilated area, out of direct sunlight, and away from incompatible materials. Store in an area that does NOT have sprinkers. Keep containers tightly closed. Store away from ignition sources. Protect from air and

FIRST AID MEASURES

moisture.

Specific Measures:

Eyes: Do not allow victim to rub eyes. Let eyes to water naturally for a few minutes. If particles/dusts are not dislodged, flush eyes thoroughly with gently running water, holding eyelids open while flushing, for five (5) minutes or until particle is removed. Do not attempt to manually removed any remaining particles. Get medical attention if irritation develops, or if particles cannot be removed by flushnia.

Skin: Wash skin with plenty of running water for five (5) minutes or until chemical is removed. If irritation persists, get medical attention.

Inhalation: Remove to fresh air. Give oxygen and get medical attention for any breathing difficulty.

Ingestion: Probably not applicable; not a normal route of exposure. If victim is alert and not convulsing, rinse mouth thoroughly with water and give 1 to 2 glasses of water to drink to dilute. If victim feels unwell, or if a very large amount has been ingested, get medical attention.

REFERENCES USED

CCINFO disc: MSDS's, January 2007 Budavari: The Merck Index, 12th ed., 1997

Sax, Lewis: Hawley's Condensed Chemical Dictionary, 11th

ed., 1987

Sax: Dangerous Properties of Industrial Materials, 5th ed.,

1979

Suppliers' Material Safety Data Sheets

ADDITIONAL INFORMATION

Date Issued: May 2, 1991 Revision: January 2007

MSDS: 9060-1

Proposed WHMIS Designation: B6 (as dust)

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